Biology 150:	1st in-class	examination
Sept 19, 2008		

Circle the lab you attend: Wednesday, 3:00-4:50, Wednesday, 9:00-10:50, Friday, 9:00-10:50

Answer the questions in the space provided and you may also use the back of the page to complete your response. There are 30 questions worth a total of 50 points (plus a couple of bonus questions [3 points total]). The point value of individual questions appears in parentheses.

- 1. There are several characteristics of all living things. A bacterium in your intestines, when suddenly surrounded by the milk you just drank, begins to express (or "turns on") the genes to produce the enzymes necessary to metabolize milk is an example of which of those characteristics? (1)
- 2. All scientific conclusions rest on three assumptions. Two of these are the assumption of natural causality and the assumption of common perception. Name the last of these assumptions and explain what it means. (2)
- 3. We can distinguish between two types of science. One of these, hypothetical (or explanatory) science is conducted using the scientific method. The other type of science is not. Name that other type of science. (1)
- 4. As an example of the scientific method I described the work of Barry Marshal and Robin Warren. State the steps of the scientific method and how it was used by Marshal and Warren. (4)

5. Darwin recognized that competition for survival exists (i.e. that more of any species or population are born than can possibly survive) and that individual organisms tend to pass on their physical characteristics to their offspring. What was his other important insight? (2)

	Which of the following vegetables was not derived from wild mustard: broccoli, cabbage, kohlrabi, cauliflower, carrot, brussel sprouts, kale? What do we call the process used to derive these vegetable over the last 2500 years? (2)
7.	What is meant by uniformitarianism? (2)
8.	What are the three most common elements in the human body? What is the fourth? (2)
	The number of what subatomic particle in an atom determines both the atom's atomic number and the number of its electrons? (1)
10.	Carbon atoms have electrons and form covalent bonds. (2)
11.	The decay of ¹⁴ C produces (1)
12.	For each of the following bonds indicate if they are non-polar covalent, ionic, or polar covalent.(1) C-C, O-H, Na-Cl, N-H, C-H
13.	If the pH is 7, what is the [H ⁺]? (Don't forget the units!) (1)
14.	Molecules that tend to stabilize pH by taking up or releasing H ⁺ are called (1)
15.	Relative to water is the specific heat of the Styrofoam used to make disposable cups higher, the same, or lower? (1)
16.	By definition, all organic molecules contain (1)
17.	Surface tension is explained by the tendency of water molecules to form bonds between one another. (1)
18.	Water climbing up a paper towel is an example of what phenomenon? (1)
19.	Molecules that do not dissolve easily in water are said to be (1)
20.	Draw the structure of the following structural groups: (4) a) Amino b) carboxyl c) phosphate d) hydroxyl

21. Name one disaccharide and name its constituent monosaccharides. (2)
22. Insect exoskeletons are made of what polysaccharide? It is formed of what monomers or subunits?(2)
23. Fats and oils are also called (1)
24. What happens during partial hydrogenation of vegetable oil? (3)
25. Describe the structure of a phospholipid. (3)
26. Cholesterol, testosterone, and estadiol are examples of what type of lipid? (1)
27. Draw the structure of an amino acid. (1)
28. What is the primary structure of a protein? (1)
29. Give two examples of the secondary structure of a protein. (2)

30. Name the two forms of nucleic acid. (2)
Bonus questions:
1. Marshal and Warren did their work in what country? (1)

2. The following two molecules are what type of isomer of each other? Draw the third. (2)

~	/ 150: 2nd in-clas r 10, 2008	ss examination	Name_	
Circle t	he lab you attend:	Wednesday, 3:00-4:50,	Wednesday, 8:00-9:50,	Friday, 8:00-9:50
respons	e. There are 23 qu	the space provided and you uestions worth a total of 5 ions appears in parenthese	ou may also use the back of t 0 points (plus 3 one point bo es.	he page to complete your onus questions). The point
1.	The sugar in each distinctive about		y carbons? Name four of the	em and describe what is
2.	RNA is single str DNA. What else	randed and DNA is double differs between RNA and	e stranded. The sugar lacks a l DNA? (1)	an OH in RNA but not in
3.	The DNA molect	ule is said to be anti-paral	lel. What does that mean? (1	1)
4.	Which type of ce	ell is typically smallest pla	ant, animal, or bacterial? Ho	w small are those cells? (2)
5.	What is the princ	ciple difference between e	ukaryotes and prokaryotes?	(1)
6.	Why are cells sm	nall? (2)		

- 7. What is chromatin composed of? (1)
- 8. Name and describe the two principle forms of the endoplasmic reticulum. What is the function of each? (2)

9.	What is the golgi? From what is it formed? What does it do? (2)
10.	What is a lysosome? What does it contain? Where is it formed? (2)
11.	What is a glyoxisome? What does it do in what organisms? (2)
12.	Name and describe two forms of endocytosis. (2)
13.	The endosymbyosis theory explains the existence of what organelles. What evidence supports the idea? (4)
14.	Name each of the three types of cytoskeletal protein fibers and name the type of protein(s) they are made of. (3)
15.	Describe the cycle of the motor protein (myosin head) as it functions during muscle contraction indicating ATP binding, hydrolysis, attachment, powerstroke, and the protein that the motor protein attaches to. (4)

	16. How wide is the cell membrane? (1)
	17. Describe (and or diagram) the fluid mosaic model. How has it been modified since originally proposed? (4)
	18. Define diffusion. What is the rate of diffusion proportional to? (2)
	19. You have a piece of closed dialysis tubing (artificial selectively permeable membrane) containing 0.2 M sucrose in a beaker containing pure water what will happen to the tubing? What is the name for the process involved? (1)
	20. Facilitated diffusion transporters are divided into two types name and describe each. Describe gating and selectivity filters and indicate to which these features apply. (2)
	21. Name and explain the difference between the two forms of cotransport. (2)
	22. What does the principle ion pump of plant, fungi, and bacterial cells transport? (1)
	23. Describe the sodium-potassium pump. What does it transport? How is it electrogenic? Where does the energy come from? (4)
В	onus questions: 1. How wide is actin? (1)
	2. What is dynein? (1)
	3 What component of animal cell membranes explains their fluidity? (1)

Biology 150: 3rd in-class examination October 31, 2008		Name	
	ne lab you attend: Wednesday, 3:00-4:50,	Wednesday, 8:00-9:50,	Friday, 8:00-9:50
response	the questions in the space provided and you e. There are 22 questions worth a total of 50 pidual questions appears in parentheses.	•	
1. V	What does the first law of thermodynamics st	ate? (1)	
2. N	What does the second law of thermodynamic	state? (1)	
	When O_2 and wood combine in combustion to neither? (1)	o produce H ₂ O and CO ₂ , is	Δ H negative, positive, or
4. V	When ice melts: (3) a) Is ΔG negative, positive, or neither?		
	b) Is ΔH negative, positive, or neither?		
	c) Is ΔS negative, positive, or neither?		
5. A	A non-spontaneous reaction is referred to as a	1 r	eaction. (1)
6. V	What is meant if a chemical reaction is said to	have reached chemical ed	quilibrium? (1)
	For a specific reaction the chemical equilibrium spontaneous b) superficial c) non-spontaneous		
a t c	Comparing an enzyme catalyzed reaction to to a) ΔG is more negative b) ΔG is more positive c) E_A is smaller d) both a and c e) none of the above	he same reaction uncatalyz	zed: (1)

9. Enzymes are said to be highly specific. What does this mean? (1)

 Describe and/or diagram how ATP hydrolysis actually functions to drive endergonic reactions in so-called "coupled reactions". (2)
11. What is meant by feedback inhibition? (1)
12. Define respiration. (1)
13. Briefly outline the glycolysis pathway. Name the starting molecule, at two intermediates, the product molecule(s) and indicate the involvement of any energy carrier molecules. (6)
14. The carbon product of glycolysis (previous question) passes into the mitochondria describe the first reaction that occurs. (3)

15.	Outline the Kreb's (citric acid) cycle. What molecule donates carbon to the pathway combining with what four carbon molecule? What six carbon molecule is produced? In a single turn of the cycle indicate the important redox reactions and any ATP produced. What is the fate of the donated carbon molecules? (4)
16.	Describe and/or diagram the process, in the presence of ample O ₂ , by which reduced electron carrier molecules produced by the Kreb's cycle are oxidized. Name those electron carrier molecules. Also indicate how this oxidation is coupled to the production of ATP. (4)
17.	From a single molecule of glucose the processes of glycolysis and aerobic respiration yields considerable ATP. How much? Give an accounting of the ATP produced. Show or describe which stages yield how much ATP. Which ATP molecules result from substrate level phosphorylation? Which from oxidative phosphorylation? (4)
18	A yeast and sugar solution not exposed to O2 produces considerable gas. What is that gas? Name and describe the process that yields this gas. What other molecule is produced in exactly the same quantity as the gas molecules? (3)

19. What is VO2 max? If an athlete significantly improved his VO2 max which would most improve his sprinting, middle distance running, or marathon running? Why? (3)
20. Describe the composition of a light harvesting complex. (1)
21. Diagram non-cyclic photophosphorylation. Show name and/or show the location of the photosystems, light absorption, electron transport, water splitting, the production of energized electron carriers, H ⁺ transport and ATP synthesis. (6)
22. What photosystem is involved in cyclic photophosphorylation? (1)

BONUS QUESTIONS:

- (1) The chemical 2,4-dinitrophenol (DNP) was sold as a diet aid beginning about 1930. Explain how it might help dieters. Could rotenone serve the same purpose, why or why not? (2)
- (2) DNP was banned as a diet aid in 1938 but, even today, is still used by some members of the body building subculture. At least three deaths have been reported in the last three years. How might DNP result in death? (1)
- (3) Consider the reaction $A + B \rightarrow C + D$. Suppose it has a Keq of 2. What would the Keq be of $C + D \rightarrow A + B$? (1)

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Biology	150:	4th	in-class	examination
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Circle the lab you attend: Wednesday, 3:00-4:50, Wednesday, 8:00-9:50, Friday, 8:00-9:50

Answer the questions in the space provided and you may also use the back of the page to complete your response. There are 22 questions worth a total of 50 points (plus 3 bonus questions). The point value of individual questions appears in parentheses.

Note: a copy of the genetic code is attached on the last page.

The so-called dark reactions of photosynthesis are also referred to as the ______ cycle after its discoverer or sometimes the ______ regular cycle after its two codiscovers. (1)
 In the first reaction of the dark reactions, CO2 is combined with ______ by the enzyme ______ to form an unstable six carbon compound that spontaneous forms two molecules of Phosphoglyceric acid (PGA). (2)
 In subsequent dark cycle reactions, PGA is phosphorylated and reduced using _____ and ____ (the two energy carrier molecules produced by the light reactions) to yield Phosphoglceraldehyde (PGAL). (2)
 DNA was first isolated and chemically characterized in 1870 by ______ . (1)
 Fred Griffith is credited with the discovery of bacterial transformation. What is bacterial transformation? Briefly, describe Griffith's experiment(s). (3)

6. In 1944, Oswald Avery, Colin Macleod, and Maclin McCarty showed that bacterial transformation, as demonstrated by Griffith required ______ but did not require_____.

(1)

7.	The role of DNA in heredity was demonstrated most unambiguously by Martha Chase and Alfred Hershey in 1952. Briefly describe their experiment(s). (4)
8.	From her experiments, what was Rosalind Franklin able to conclude about the structure of DNA? (3)
9.	What were/are "Chargaff's rules"? (1)
10	Describe and/or diagram eukaryotic DNA replication. Mention the role of each major enzyme complex (e.g. DNA polymerase I, DNA ligase, DNA helicase, topoisomerase, RNA primase, single strand binding proteins), okazaki fragments, and the differences between leading and lagging strand synthesis. (6)

11. Define gene. (1)
12. Define codon. (1)
13. Describe transcription. What enzyme complex performs the activity. Where does it bind? Which way does it travel? Where does transcription actually start? Where/how does transcription terminate? What is produced? (5)
14. In eukaryotes pre-mRNAs are processed to become mRNAs. Describe the changes that occur. (3)
15. What are snRNAs? (1)
16. Assume the following sequence represents one end of a mRNA (shown 5' to 3'):
methyl-GGAAGGAGGUAACACAUGCCUUCCUUAACUGCGGAGGAUAAA
a) list the first 9 amino acids that would appear in the resulting peptide (3)
b) list the anti-codons, in order, of the first 6 tRNAs involved in the synthesis of that peptide (3)
c) give the base sequence of the gene transcribed into the mRNA (3)

17	Describe and/or diagram the events of translation. In what order do the components come
17.	together? How is protein amino acid sequence determined? How is translation terminated? (5)
10	What is a frameshift mutation? What kind(s) of DNA changes produce these types of
10.	mutations?(1)
	1:00 0 (1)
19.	How do missense and nonsense mutations differ? (1)
20.	Describe and/or diagram the lac operon. Name and indicate the relative location of the different elements. Describe how changing lactose concentration alters function. (4)
21	. How do inducible and repressible operons differ? (1)
22	. Describe and/or diagram the cell cycle. Show the relative size and location of Interphase, G2, G1
	S, and Mphase. (1)

BONUS QUESTIONS:

1. Describe the experimental method used in the late 1940's in a laboratory at UC Berkeley to discover the sequence of reactions of the "dark reactions" of photosynthesis. (2)

2. Describe the experimental method used in the early 1960's in laboratories at the NIH in Maryland and at the U. of Wisconsin to discover the genetic code. (2)

3. How many introns occur in the human B-globin gene? (1)

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Biology 150: Final examination

Dec 15, 2008

Circle the lab you attend: Wednesday, 3:00-4:50,

Wednesday, 8:00-9:50,

Name

Friday, 8:00-9:50

Answer the questions in the space provided and you may also use the back of the page to complete your response. There are 46 questions worth a total of 100 points (plus 5 possible bonus points). The point value of individual questions appears in parentheses.

- 1. Mitosis begins with prophase. Name the remaining phases of mitosis and briefly indicate what occurs in each. (2)
- 2. Cytokinesis takes to possible forms. Briefly describe how it occurs in animal cells and how it occurs in plants. (2)
- 3. What is a phragmoplast? (1)
- 4. Describe binary fission. In what organisms does it occur? (2)
- 5. What is the term used to name an organized illustration or photograph of the condensed chromosomes of an individual organism in which the homologous pairs are arranged together? (1)
- 6. Humans have how many pairs of homologous chromosomes? How many are autosomes? (2)
- 7. There are three lifecycle types in nature. Name and describe(and/or illustrate) each indicating when cells are haploid and diploid, where mitosis, meiosis, and fertilization occur. (6)

8.	How does prophase I differ from prophase? (3)
9.	The genetic material you produce in the individual gametes you produce is always a mix of the genetic material you received as a zygote from your two parents. This mixing of your parental genes in your gametes is ensured by two mechanisms. Name each of these mechanisms and explain how each contributes to genetic mixing. (4)
10.	What term is used to name the location of a gene on one or both homologous chromosomes? (1)
11.	Distinguish between the meaning of phenotype and genotype. (2)
12.	What does it means to say that certain genes are tightly linked? (1)
13.	Gregor Mendel did his important experiments in what decade (e.g. the 1830s the 1920s etc)? (1)
14.	In Mendel's monohybrid cross involving flower color all the F_1 individuals looked the same. What was their phenotype and what was their genotype(s)? (2)
15.	Allowed to self-pollinate, the F_1 generation produced a mixture of phenotypes and genotypes in the F_2 generation. What were their phenotypes and what was their genotypes? (2)
16	In a cross between Aa and aa, what phenotypes and genotypes would result in what proportions? (1)
17	In a cross between AaBbCCDDEE and AaBbccddee, what proportion of the progeny would have the dominant phenotype for all five genes? (1)
18	. Distinguish between sexual recombination and genetic recombination. (2)

19.	You experimentally cross a true-breeding fruit fly with both forked bristles and vermillion eyes with a wild type fly. All the progeny (i.e. the F1) have normal (i.e. wild type) bristles and eyes. You allow the progeny to interbreed. The resulting F2 consist mostly of a 3:1 ratio of flies with normal bristles and eyes to flies with both forked bristles and vermillion eyes with a few with either forked bristles and normal eyes or normal bristles and vermillion eyes. (6)
	a) What do the results tell you?
	b) Is sexual recombination evident? If so which flies form the sexual recombinant class(es)? (2)
	c) Is genetic recombination evident? If so which flies form the genetic recombinant class(es)?
20	You perform a second experiment with the flies from the previous question. You cross some of the F1 with more true-breeding flies with forked bristles and vermillion eyes. The result in the progeny is a total of 1000 flies: 400 with normal bristles and eyes, 400 with forked bristles and vermillion eyes, 100 with forked bristles and normal eyes and 100 with normal bristles and vermillion eyes. How far apart are the genes? (2)
21	You have an interest in a certain wildflower and have collect a few mutant varieties you have found in nature. You already have some plants that lack the normal flower color and recently you have discovered a second population. Crossing plants for either population with the normal wild type plants shows the white flower phenotype to be recessive. Curious to know more about the genetic basis of the two white flowered populations you cross two white flowered plants from the two different populations the result is plants that have normal flower color. Explain the results. (2)
22	There are several characteristics of all living things. A bacterium in your intestines, when suddenly surrounded by the milk you just drank, begins to express (or "turns on") the genes to produce the enzymes necessary to metabolize milk is an example of which of those characteristics? (1)

23.	As an example of the State the steps of the	he scientific method ne scientific method	d I described the value of the	work of Barry Mased by Marshal a	arshal and Robin and Warren. (4)	Warren.
24.	Darwin recognized are born than can p characteristics to the	l that competition foossibly survive) an heir offspring. Wha	d that individual	organisms tend to	pass on their pl	opulation nysical
25.	For each of the fol	lowing bonds indic	ate if they are nor _, Na-Cl	n-polar covalent,	ionic, or polar co	ovalent.(1
26.	If the pH is 7, wha	t is the [H ⁺]? (Don'	t forget the units!)(1)		
27.	Molecules that ten	d to stabilize pH by	taking up or rele	asing H ⁺ are calle	ed	. (1)
28.	Draw the structure a) Amino		ructural groups: (c) phosph		d) hydroxyl	
29	. The sugar in each distinctive about e	nucleotide has how ach. (4)	many carbons? I	Name four of ther	n and describe w	/hat is

30. RNA is single stranded and DNA is double strand DNA. What else differs between RNA and DNA?	
31. The DNA molecule is said to be anti-parallel. Wh	nat does that mean? (1)
32. Name and describe the two principle forms of the each? (2)	endoplasmic reticulum. What is the function of
33. What is the golgi? From what is it formed? What	does it do? (2)
34. What is a lysosome? What does it contain? When	e is it formed? (2)
35. What is a glyoxisome? What does it do in what o	organisms? (2)
36. You have a piece of closed dialysis tubing (artification). 2 M sucrose in a beaker containing pure water for the process involved? (1)	cial selectively permeable membrane) containing what will happen to the tubing? What is the name
37. What does the first law of thermodynamics state	? (1)
38. What does the second law of thermodynamic sta	te? (1)
39. When O ₂ and wood combine in combustion to property neither? (1)	roduce H ₂ O and CO ₂ , is ΔH negative, positive, or

40.	Briefly outline the glycolysis pathway. Name the starting molecule, at two intermediates, the product molecule(s) and indicate the involvement of any energy carrier molecules. (6)
41.	The carbon product of glycolysis (previous question) passes into the mitochondria describe the first reaction that occurs. (3)
42.	Outline the Kreb's (citric acid) cycle. What molecule donates carbon to the pathway combining with what four carbon molecule? What six carbon molecule is produced? In a single turn of the cycle indicate the important redox reactions and any ATP produced. What is the fate of the donated carbon molecules? (4)
43	. In 1944, Oswald Avery, Colin Macleod, and Maclin McCarty showed that bacterial transformation, as demonstrated by Griffith required but did not require
	(1)

44.	4. The role of DNA in heredity was demonstrated most unambiguously Hershey in 1952. Briefly describe their experiment(s). (4)	by Martha Chase and Alfred
45.	5. Define codon. (1)	
46.	6. Describe transcription. What enzyme complex performs the activity. way does it travel? Where does transcription actually start? Where/hoterminate? What is produced? (5)	
RONU	US QUESTION:	
1.	. The following scientists all received the Noble prize. Match the prize accomplishment for which he was awarded the prize (5):	winners with the
		etic code
	Peter Michel the lac	operon
		ıble helix
		ic acid cycle
	Marshal Nirenburgchemic	osmosis)